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L1 0 1933:563/RAN.CAPLUS

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L2 45013 MACDONALD?/RAU

=> s 1932/RPY  
L3 33663 1932/RPY

=> s 1933/RPY  
L4 30055 1933/RPY

=> s l2 and l3  
L5 377 L2 AND L3

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L9 ANSWER 1 OF 1 CAPLUS COPYRIGHT 2009 ACS on STN  
ACCESSION NUMBER: 1998:763808 CAPLUS  
DOCUMENT NUMBER: 130:117157

ENTRY DATE: Entered STN: 07 Dec 1998  
TITLE: NH radical reactions  
AUTHOR(S): Hack, W.  
CORPORATE SOURCE: Max-Planck-Institut fur Stromungsforschung, Gottingen,  
D-37073, Germany  
SOURCE: N-Centered Radicals (1998), 413-466. Editor(s):  
Alfassi, Zeev B. Wiley: Chichester, UK.  
CODEN: 67ABAQ  
DOCUMENT TYPE: Conference; General Review  
LANGUAGE: English  
CLASSIFICATION: 74-0 (Radiation Chemistry, Photochemistry, and  
Photographic and Other Reprographic Processes)  
Section cross-reference(s): 78

ABSTRACT:  
A review with 441 refs. in which reactivity of various electronic states of imidogen radical NH are discussed, in particular electronic ground state X3Σ- and the first electronically excited singlet state a1Δ. Production and detection of ground state- and excited NH radicals is discussed, along with quenching of excited NH and elementary reactions of these radicals. The reactions included reactions in the N-H system (with H, H2, N, N2, NH, NH3, N2H4), in the N-H-O system (with O, NO, H2O, O2, H2O2, OH), in the N-H-C(Si) system (with SiH4, saturated hydrocarbons, unsatd. hydrocarbons), in the N-H-C-O system (with CO, CO2, MeOH), in the N-H-halogen system (with HF, HCl, HCN, F2, Cl2, ClF).

SUPPL. TERM: review imidogen radical reaction ground state excited;  
photolysis imidogen radical reaction review  
INDEX TERM: Reaction  
Reaction kinetics  
(reactions of imidogen radical in ground state)  
INDEX TERM: Electronic excitation  
Electronic transition  
Photochemistry  
Photolysis  
Photolysis kinetics  
(reactions of imidogen radical in ground state and  
electronically excited states)  
INDEX TERM: Alkanes, reactions  
Alkenes, reactions  
ROLE: PEP (Physical, engineering or chemical process); RCT  
(Reactant); PROC (Process); RACT (Reactant or reagent)  
(reactions of imidogen radical in ground state and  
electronically excited states)  
INDEX TERM: 67-56-1, Methanol, reactions 74-90-8, Hydrogen cyanide,  
reactions 124-38-9, Carbon dioxide, reactions 302-01-2,  
Hydrazine, reactions 630-08-0, Carbon monoxide, reactions  
3352-57-6, Hydroxyl, reactions 7647-01-0, Hydrogen  
chloride, reactions 7664-39-3, Hydrogen fluoride,  
reactions 7664-41-7, Ammonia, reactions 7722-84-1,  
Hydrogen peroxide, reactions 7727-37-9, Nitrogen,  
reactions 7782-41-4, Fluorine, reactions 7782-44-7,  
Oxygen, reactions 7782-50-5, Chlorine, reactions  
7790-89-8, Chlorine fluoride (ClF) 7803-62-5, Silane,  
reactions 10102-43-9, Nitrogen oxide (NO), reactions  
12385-13-6, Hydrogen atom, reactions 13774-92-0, Imidogen  
17778-80-2, Oxygen atom, reactions 17778-88-0, Nitrogen  
atom, reactions  
ROLE: PEP (Physical, engineering or chemical process); RCT  
(Reactant); PROC (Process); RACT (Reactant or reagent)  
(reactions of imidogen radical in ground state and  
electronically excited states)  
REFERENCE COUNT: 443 THERE ARE 443 CITED REFERENCES AVAILABLE FOR THIS  
RECORD.

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      596 MACDONALD?/RIN
      43866 MCDONALD?/RAU
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      451 MAC DONALD?/RAU
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L15 ANSWER 1 OF 1 CAPLUS COPYRIGHT 2009 ACS on STN  
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ABSTRACT:

A review. The environmental fate of sodium azide ( $\text{NaN}_3$ ) is of considerable interest, given the recent surge in production to satisfy demand for automobile air bag inflators, where it serves as the principal active ingredient. Since the mid-1990s, demand for sodium azide has exceeded 5 million kg/yr and most passenger vehicles sold in the USA now contain approx. 300 g ( $\approx 0.7$  lb) of sodium azide. This has greatly increased the potential for accidental environmental releases and for human exposure to this highly toxic, broad-spectrum biocide. It can be argued that a new environmental threat has developed because not only are millions of kilograms of sodium azide now transported to and processed at air bag inflator factories, but also this substance is now widely distributed throughout the developed world in automobiles. Even if sodium azide were to be replaced by a more benign propellant in the future, the problem of safely disposing of large quantities of azide will remain as the vehicle fleet ages and is retired to scrap yards and shredders. Unfortunately, the environmental fate of sodium azide is unknown, so it is difficult to effectively manage releases. The problem is compounded by the fact that aqueous sodium azide is readily hydrolyzed to yield hydrazoic acid ( $\text{HN}_3$ ), a volatile substance that partitions strongly to the gas phase.

SUPPL. TERM: review sodium azide automobile airbag derived environmental  
fate  
INDEX TERM: Automobiles  
Vehicles  
(airbags; environmental fate of sodium azide derived from  
vehicle airbags)  
INDEX TERM: Air pollution  
Airbags (protective)  
Environmental pollution  
Human  
Toxicity

(environmental fate of sodium azide derived from  
automobile airbags)

INDEX TERM: 26628-22-8, Sodium azide  
ROLE: POL (Pollutant); TEM (Technical or engineered material  
use); OCCU (Occurrence); USES (Uses)  
(environmental fate of sodium azide derived from  
automobile airbags)

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MISSING OPERATOR DONALD?), 1932?/RE

The search profile that was entered contains terms or  
nested terms that are not separated by a logical operator.

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=> S E1

L17          1 "1915:10274"/AN

=> D L17 BIB,ABS

L17  ANSWER 1 OF 1 CAPLUS COPYRIGHT 2009 ACS on STN
AN  1915:10274 CAPLUS
DN  9:10274
OREF 9:1586d-e
TI  Behavior of certain hydrazine salts on decomposition by heat
AU  Turrentine, J. W.
CS  Cornell Univ.
SO  Journal of the American Chemical Society (1915), 37, 1105-14
    CODEN: JACSAT; ISSN: 0002-7863
DT  Journal
LA  Unavailable
AB  The decompose of hydrazine monochlorate by heat in neutral aqueous solns.
    yielded no HN3; HN3 was formed, however, in H2SO4 solution. The corresponding
    di-salt yielded HN3 both in the presence and absence of H2O4. The mono-
    and diperchlorate of N2H4 yielded no HN3 either in acidified or
    unacidified solns. When the dry crystallized salt was decomposed by heat in a
    current of CO2, HN3 was produced from the latter but not from the former.
    Both yielded gaseous Cl, N and O, and Cl and OH ions. The mono- and
    disulfates, under similar conditions, gave off H2O, H2S, SO2 and S, but no
    HN3. HN3 was not oxidized on boiling with H2O2 in alkaline solution; it may be
    separated from sulfites in this way, and this principle has been applied in a
    method (described) for detecting HN3 in presence of SO2.

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L18          1 1915:10274/RAN.CAPLUS

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L18  ANSWER 1 OF 1 CAPLUS COPYRIGHT 2009 ACS on STN
ACCESSION NUMBER: 2003:959696 CAPLUS
DOCUMENT NUMBER: 140:257580
ENTRY DATE: Entered STN: 09 Dec 2003
TITLE: Environmental fate of sodium azide derived from
       automobile airbags
AUTHOR(S): Betterton, Eric A.
CORPORATE SOURCE: Department of Atmospheric Sciences, University of
                   Arizona, Tucson, AZ, 85721-0081, USA
SOURCE: Critical Reviews in Environmental Science and
        Technology (2003), 33(4), 423-458
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CODEN: CRETEK; ISSN: 1064-3389  
PUBLISHER: CRC Press LLC  
DOCUMENT TYPE: Journal; General Review  
LANGUAGE: English  
CLASSIFICATION: 59-0 (Air Pollution and Industrial Hygiene)  
ABSTRACT:  
A review. The environmental fate of sodium azide (NaN<sub>3</sub>) is of considerable interest, given the recent surge in production to satisfy demand for automobile air bag inflators, where it serves as the principal active ingredient. Since the mid-1990s, demand for sodium azide has exceeded 5 million kg/yr and most passenger vehicles sold in the USA now contain approx. 300 g ( $\approx$ 0.7 lb) of sodium azide. This has greatly increased the potential for accidental environmental releases and for human exposure to this highly toxic, broad-spectrum biocide. It can be argued that a new environmental threat has developed because not only are millions of kilograms of sodium azide now transported to and processed at air bag inflator factories, but also this substance is now widely distributed throughout the developed world in automobiles. Even if sodium azide were to be replaced by a more benign propellant in the future, the problem of safely disposing of large quantities of azide will remain as the vehicle fleet ages and is retired to scrap yards and shredders. Unfortunately, the environmental fate of sodium azide is unknown, so it is difficult to effectively manage releases. The problem is compounded by the fact that aqueous sodium azide is readily hydrolyzed to yield hydrazoic acid (HN<sub>3</sub>), a volatile substance that partitions strongly to the gas phase.

SUPPL. TERM: review sodium azide automobile airbag derived environmental fate  
INDEX TERM: Automobiles  
Vehicles  
(airbags; environmental fate of sodium azide derived from vehicle airbags)  
INDEX TERM: Air pollution  
Airbags (protective)  
Environmental pollution  
Human  
Toxicity  
(environmental fate of sodium azide derived from automobile airbags)  
INDEX TERM: 26628-22-8, Sodium azide  
ROLE: POL (Pollutant); TEM (Technical or engineered material use); OCCU (Occurrence); USES (Uses)  
(environmental fate of sodium azide derived from automobile airbags)  
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=> s macdonald?/rau  
L19 45013 MACDONALD?/RAU

=> s macdonald/rau  
L20 557 MACDONALD/RAU

=> s 1932/RPY  
L21 33663 1932/RPY

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L22 9 L19 (L) L21

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YOU HAVE REQUESTED DATA FROM 9 ANSWERS - CONTINUE? Y/(N):y

L22 ANSWER 1 OF 9 CAPLUS COPYRIGHT 2009 ACS on STN  
 ACCESSION NUMBER: 2008:645866 CAPLUS  
 DOCUMENT NUMBER: 148:555843  
 ENTRY DATE: Entered STN: 30 May 2008  
 TITLE: Bubbles, gating, and anesthetics in ion channels  
 AUTHOR(S): Roth, Roland; Gillespie, Dirk; Nonner, Wolfgang;  
 Eisenberg, Robert E.  
 CORPORATE SOURCE: Max-Planck Institut fuer Metallforschung, Stuttgart,  
 Germany  
 SOURCE: Biophysical Journal (2008), 94(11), 4282-4298  
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 CLASSIFICATION: 6-7 (General Biochemistry)

**ABSTRACT:**

We suggest that bubbles are the bistable hydrophobic gates responsible for the on-off transitions of single channel currents. In this view, many types of channels gate by the same phys. mechanism-dewetting by capillary evaporation-but different types of channels use different sensors to modulate hydrophobic properties of the channel wall and thereby trigger and control bubbles and gating. Spontaneous emptying of channels has been seen in many simulations. Because of the physics involved, such phase transitions are inherently sensitive, unstable threshold phenomena that are difficult to simulate reproducibly and thus convincingly. We present a thermodyn. anal. of a bubble gate using morphometric d. functional theory of classical (not quantum) mechanics. Thermodyn. anal. of phase transitions is generally more reproducible and less sensitive to details than simulations. Anesthetic actions of inert gases - and their interactions with hydrostatic pressure (e.g., nitrogen narcosis) - can be easily understood by actions on bubbles. A general theory of gas anesthesia may involve bubbles in channels. Only expts. can show whether, when, or which channels actually use bubbles as hydrophobic gates: direct observation of bubbles in channels is needed. Existing expts. show thin gas layers on hydrophobic surfaces in water and suggest that bubbles nearly exist in bulk water.

SUPPL. TERM: ion channel microcapillary bubble gating switch noble gas anesthetic  
INDEX TERM: Anesthetics  
(bubbles-dependent gating model for ion channels suggests mechanism for hydrostatic pressure-associated anesthetic action of inert gases)  
INDEX TERM: Noble gases, biological studies  
ROLE: BSU (Biological study, unclassified); BIOL (Biological study)  
(bubbles-dependent gating model for ion channels suggests mechanism for hydrostatic pressure-associated anesthetic action of inert gases)  
INDEX TERM: Capillary tubes  
Molecular switches  
(bubbles-dependent gating model suggests ion channels exhibit microcapillary-like behavior by switching between water-filled conductive state and vapor-locked nonconductive state)  
INDEX TERM: Ion channels  
ROLE: BSU (Biological study, unclassified); BIOL (Biological study)  
(bubbles-dependent gating model suggests ion channels exhibit microcapillary-like behavior by switching between water-filled conductive state and vapor-locked nonconductive state)  
INDEX TERM: Wetting  
(dewetting; bubbles-dependent gating model suggests ion channels exhibit microcapillary-like behavior by switching between water-filled conductive state and vapor-locked nonconductive state)  
INDEX TERM: Pressure  
(hydrostatic; bubbles-dependent gating model for ion channels suggests mechanism for hydrostatic pressure-associated anesthetic action of inert gases)  
INDEX TERM: Bubbles  
(microbubbles; bubbles-dependent gating model suggests ion channels exhibit microcapillary-like behavior by switching between water-filled conductive state and vapor-locked nonconductive state)  
INDEX TERM: 7732-18-5, Water, biological studies  
ROLE: BSU (Biological study, unclassified); BIOL (Biological study)

(bubbles-dependent gating model suggests ion channels exhibit microcapillary-like behavior by switching between water-filled conductive state and vapor-locked nonconductive state)

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L22 ANSWER 2 OF 9 CAPLUS COPYRIGHT 2009 ACS on STN  
 ACCESSION NUMBER: 2007:794911 CAPLUS  
 DOCUMENT NUMBER: 147:426264  
 ENTRY DATE: Entered STN: 22 Jul 2007  
 TITLE: Ascorbic acid  
 AUTHOR(S): Johnston, Carol S.; Steinberg, Francene M.; Rucker, Robert B.  
 CORPORATE SOURCE: Department of Nutrition, Arizona State University, Mesa, AZ, USA  
 SOURCE: Handbook of Vitamins (4th Edition) (2007), 489-520.

Editor(s): Zempleni, Janos. CRC Press LLC: Boca Raton, Fla.  
CODEN: 69JLTW; ISBN: 978-0-8493-4022-2

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LANGUAGE: English  
CLASSIFICATION: 18-0 (Animal Nutrition)  
ABSTRACT:  
A review on ascorbic acid (vitamin C), including its chemical and food sources. The history of ascorbic acid is discussed, along with its nomenclature and structure, phys. and chemical properties, chemical and biol. synthesis, biochem. functions, relationship with glutathione, role as an antioxidant, link with gene expression, and metabolism and regulation. The selected clin. features important to ascorbic acid status are examined, together with progression of selected chronic diseases, ascorbic acid requirements, allowances, and upper limits, iron-related disorders, and vitamin B12.

SUPPL. TERM: review ascorbic acid antioxidant chronic disease nutrition  
INDEX TERM: Disease, animal  
(chronic; poor ascorbic acid status was related to atherosclerosis, cataract, cancer and worsening of respiratory function in smoker and elderly patient)  
INDEX TERM: Aging, animal  
(elderly; poor ascorbic acid status was related to atherosclerosis, cataract and worsening of respiratory function in smoker and elderly patient)  
INDEX TERM: Antioxidants  
Atherosclerosis  
Cataract  
Human  
Metabolism, animal  
Nutrition, animal  
(poor ascorbic acid status was related to atherosclerosis, cataract and worsening of respiratory function in smoker and elderly patient)  
INDEX TERM: Neoplasm  
(poor ascorbic acid status was related to atherosclerosis, cataract, cancer and worsening of respiratory function in smoker and elderly patient)  
INDEX TERM: Scurvy  
(poor ascorbic acid status was related to atherosclerosis, scurvy, cataract and worsening of respiratory function in smoker and elderly patient)  
INDEX TERM: 50-81-7, Ascorbic acid, biological studies 70-18-8, Glutathione, biological studies  
ROLE: BSU (Biological study, unclassified); BIOL (Biological study)  
(poor ascorbic acid status was related to atherosclerosis, cataract and worsening of respiratory function in smoker and elderly patient)  
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 ABSTRACT:  
 A review. Originally developed in the late 1960s, anodic bonding, also known as electrostatic bonding, field-assisted bonding or Mallory bonding, has become one of the most important silicon packaging techniques. Despite its industrial relevance the bonding mechanism is mainly only qual. understood and is almost solely applied to the bonding of silicon to Pyrex glass. The objective of the present paper is to review the current state of knowledge of the anodic bonding process. Possible material combinations and current scientific and industrial applications of this bonding technique are reviewed. The various aspects of the bonding process, such as the creation of intimate contact, the cation movement in the glass and the interfacial chemical reactions, are discussed in

detail and related to the external current measured during bonding to describe the bonding process quant. A better understanding of the process itself should help not only to improve the process control and the quality of devices, but also to broaden the application of this successful bonding technique to more challenging designs, to smaller device sizes and to systems other than silicon-Pyrex glass.

SUPPL. TERM: review anodic bonding silicon glass packaging  
INDEX TERM: Diffusion  
Electric current  
Electronic packaging process  
Glass substrates  
Interfacial reaction  
Joining  
    (anodic bonding)  
INDEX TERM: 7440-21-3, Silicon, processes  
ROLE: PEP (Physical, engineering or chemical process); TEM (Technical or engineered material use); PROC (Process); USES (Uses)  
    (anodic bonding)  
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L22 ANSWER 4 OF 9 CAPLUS COPYRIGHT 2009 ACS on STN  
 ACCESSION NUMBER: 2004:142631 CAPLUS  
 DOCUMENT NUMBER: 140:159044  
 ENTRY DATE: Entered STN: 22 Feb 2004  
 TITLE: Mold-resistant corrugated cardboard for void-forming structures  
 INVENTOR(S): Comeau, Dominic Hamel; Rosenblat, Michael L.  
 PATENT ASSIGNEE(S): Voidform International Ltd., Can.  
 SOURCE: U.S. Pat. Appl. Publ., 8 pp.  
 CODEN: USXXCO  
 DOCUMENT TYPE: Patent  
 LANGUAGE: English  
 INT. PATENT CLASSIF.:

MAIN: B32B001-00  
 US PATENT CLASSIF.: 428178000  
 CLASSIFICATION: 5-2 (Agrochemical Bioregulators)  
 Section cross-reference(s): 43  
 FAMILY ACC. NUM. COUNT: 1  
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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US 6794017	B2	20040921		
CA 2431614	A1	20031212	CA 2003-2431614	20030610
PRIORITY APPLN. INFO.:			US 2002-388134P	P 20020612

PATENT CLASSIFICATION CODES:

PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
US 20040033343	ICM	B32B001-00
	INCL	428178000
	IPCI	B32B0001-00 [ICM, 7]
	IPCR	B32B0001-00 [I,C*]; B32B0001-00 [I,A]; B32B0003-28 [I,C*]; B32B0003-28 [I,A]; D21H0011-00 [I,C*]; D21H0011-00 [I,A]; E02D0031-00 [I,C*]; E02D0031-10 [I,A]; E04B0005-32 [I,C*]; E04B0005-32 [I,A]; E04C0003-30 [I,C*]; E04C0003-34 [I,A]; E04G0015-00 [I,C*]; E04G0015-06 [I,A]; B31F0001-20 [I,C*]; B31F0001-28 [I,A]; B32B0027-08 [I,A]; D21H0017-00 [N,C*]; D21H0017-28 [N,A]; D21H0021-14 [N,C*]; D21H0021-36 [N,A]; D21H0027-30 [I,C*]; D21H0027-40 [I,A]
	NCL	428/178.000; 428/182.000; 156/292.000; 162/161.000
	ECLA	B32B027/08; B31F001/28B1; D21H027/40; E02D031/10; E04B005/32; E04C003/34; E04G015/06B; N21H; N21H
CA 2431614	IPCI	D21H0021-36 [ICM, 7]; D21H0021-14 [ICM, 7,C*]; D21F0011-12 [ICS, 7]; D21F0011-00 [ICS, 7,C*]; B31F0001-20 [ICS, 7]
	IPCR	B31F0001-20 [I,C*]; B31F0001-20 [I,A]; D21F0011-00 [I,C*]; D21F0011-12 [I,A]; D21H0021-14 [I,C*]; D21H0021-36 [I,A]

ABSTRACT:

Corrugated cardboard that resists the growth of mold includes single and double face liners produced with pulp and starch, where the latter is diluted with water to which biocides have been added. The liner biocides include 5-chloro-2-methyl-4-isothiazolin-3-one, hypochlorite and sodium hydroxide and sodium bromide. A fluted medium is sandwiched between the single and double face liners with an adhesive that is produced with diluted starch and to which biocides have also been added. The biocides in the adhesive include 1,2-benzothiazol-3(2H)-one and poly[oxyethylene(dimethylimino)ethylene dichloride].

SUPPL. TERM:	mold corrugated cardboard fungicide
INDEX TERM:	Paperboard (corrugated; mold-resistant corrugated cardboard for void-forming structures)
INDEX TERM:	Biocides (in mold-resistant corrugated cardboard for void-forming structures)
INDEX TERM:	Fungicides Mold (fungus) (mold-resistant corrugated cardboard for void-forming structures)
INDEX TERM:	2634-33-5, 1,2-Benzisothiazol-3(2H)-one 7647-15-6, Sodium bromide., biological studies 14380-61-1, Hypochlorite 26172-55-4, 5-Chloro-2-methyl-4-isothiazolin-3-one

31512-74-0  
ROLE: BUU (Biological use, unclassified); BIOL (Biological study); USES (Uses)  
(in mold-resistant corrugated cardboard for void-forming structures)

REFERENCE COUNT: 10 THERE ARE 10 CITED REFERENCES AVAILABLE FOR THIS RECORD.

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L22 ANSWER 5 OF 9 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2002:550899 CAPLUS

DOCUMENT NUMBER: 137:89509

ENTRY DATE: Entered STN: 25 Jul 2002

TITLE: Saturated halogenated aliphatic hydrocarbons: two to four carbons

AUTHOR(S): Reid, Jon B.

CORPORATE SOURCE: University of Cincinnati, Cincinnati, OH, USA

SOURCE: Patty's Toxicology (5th Edition) (2001), Volume 5, 99-204. Editor(s): Bingham, Eula; Cohrssen, Barbara; Powell, Charles H. John Wiley & Sons, Inc.: New York, N. Y.

CODEN: 69CWST; ISBN: 0-471-31943-0

DOCUMENT TYPE: Conference; General Review

LANGUAGE: English

CLASSIFICATION: 4-0 (Toxicology)

ABSTRACT:

A review. Properties and toxicol. of saturated halogenated aliphatic hydrocarbons with two to four carbons are comprehensively surveyed.

SUPPL. TERM: review toxicol halogenated hydrocarbon

INDEX TERM: Hydrocarbons, biological studies

ROLE: BSU (Biological study, unclassified); BIOL (Biological study)

(halo, C2-4; properties and toxicol. of saturated halogenated aliphatic hydrocarbons with two to four carbons)

REFERENCE COUNT: 398 THERE ARE 398 CITED REFERENCES AVAILABLE FOR THIS RECORD.

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L22 ANSWER 6 OF 9 CAPLUS COPYRIGHT 2009 ACS on STN  
ACCESSION NUMBER: 2001:503511 CAPLUS  
ENTRY DATE: Entered STN: 12 Jul 2001  
TITLE: Assessing the Evidence for Remodelling of the Airway  
in Asthma  
AUTHOR(S): Wilson, John W.; Bamford, Tiffany L.  
CORPORATE SOURCE: Department of Respiratory Medicine, Monash Medical  
School and The Alfred Hospital, Prahran, 3181,  
Australia  
SOURCE: Pulmonary Pharmacology & Therapeutics (2001), 14(3),  
229-247  
CODEN: PPTHFJ; ISSN: 1094-5539  
PUBLISHER: Academic Press  
DOCUMENT TYPE: Journal  
LANGUAGE: English  
ABSTRACT:  
Asthma is now described as being characterized by reversible airflow  
obstruction, with bronchial inflammation and tissue remodelling of the airway  
wall. The description of remodelling has been usefully invoked to account for  
a component of airflow obstruction that is unresponsive to usual bronchodilator  
therapy. It is crucial to examine critically the evidence for this view,  
particularly the quantitation of specific changes in the epithelium, mucus  
glands, cell infiltrate, collagen, vessels and smooth muscle of the bronchial  
wall. The useful tools of immunohistochem. and mol. biol. combined with airway  
biopsy and well-designed clin. trials will be essential to determine the specific  
roles of cells and cytokines in airway remodelling in asthma. (C) 2001  
Academic Press. (c) 2001 Academic Press.  
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L22 ANSWER 7 OF 9 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2000:636245 CAPLUS  
 ENTRY DATE: Entered STN: 13 Sep 2000  
 TITLE: Apparatus for forming voids under concrete floors  
 INVENTOR(S): Rosenblat, Michael L.; Isles, Timothy J.; Dufault, Louis C.  
 PATENT ASSIGNEE(S): Void Form International Ltd., Can.  
 SOURCE: U.S., 9 pp.  
 CODEN: USXXAM  
 DOCUMENT TYPE: Patent  
 LANGUAGE: English  
 INT. PATENT CLASSIF.:  
     MAIN: B28B007-28  
     SECONDARY: B28B007-34  
 US PATENT CLASSIF.: 249184000; 249DIG2; 249061000; 249185000; 052792100;  
                       052795100; 493964000; 493966000; 493390000; 229120170  
 FAMILY ACC. NUM. COUNT: 1  
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 6116568	A	20000912	US 1998-70302	19980430
PRIORITY APPLN. INFO.:			US 1998-71974P	P 19980120
PATENT CLASSIFICATION CODES:				
PATENT NO.	CLASS	PATENT FAMILY	CLASSIFICATION CODES	
US 6116568	ICM	B28B007-28		
	ICS	B28B007-34		
	INCL	249184000; 249DIG2; 249061000; 249185000; 052792100; 052795100; 493964000; 493966000; 493390000; 229120170		
	IPCI	B28B0007-28 [ICM, 7]; B28B0007-34 [ICS, 7]		
	IPCR	B28B0007-28 [I,C*]; B28B0007-28 [I,A]; E02D0027-01 [I,C*]; E02D0027-01 [I,A]; E04G0009-08 [I,C*]; E04G0009-08 [I,A]; E04G0009-10 [I,C*]; E04G0009-10 [I,A]		
	NCL	249/184.000; 052/792.100; 052/795.100; 229/120.170; 229/120.250; 229/120.360; 249/061.000; 249/185.000; 249/DIG.002; 493/390.000; 493/964.000; 493/966.000		
	ECLA	B28B007/28; E02D027/01; E04G009/08; E04G009/10		

ABSTRACT:

A reinforced box-like structure for forming a void area in a concrete

formation. The structure includes a bottom panel having a plurality of spaced apart, parallel base partitions and a top panel having a plurality of spaced apart, parallel top partitions. The base partitions extend across the bottom panel and in a direction normal to the bottom panel, and the top partitions extend across the top panel and in a direction normal to the top panel. Each of the base partitions and top partitions include a plurality of double-thickness walls with a plurality of slots therein and flat top edges between the slots. The structure is assembled by superimposing the top panel over the bottom panel such that the base partitions extend transversely relative to the top partitions. Thus, the slots in the base partitions register in interlocking relationship with the corresponding slots in the top partitions, wherein the base partitions extend substantially to the top panel and the top partitions extend substantially to the bottom panel to create a cellular box-like structure that generally fills a space between the bottom panel and top panel with the flat top edges engaging the respective top or bottom partition.

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L22 ANSWER 8 OF 9 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1999:505202 CAPLUS  
DOCUMENT NUMBER: 131:291084  
ENTRY DATE: Entered STN: 16 Aug 1999  
TITLE: Metal-Based X-ray Contrast Media  
AUTHOR(S): Yu, Shi-Bao; Watson, Alan D.  
CORPORATE SOURCE: Torsten Almen Research Center, Nycomed Amersham Imaging, Wayne, PA, 19087, USA  
SOURCE: Chemical Reviews (Washington, D. C.) (1999), 99(9), 2353-2377  
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ABSTRACT:

A review with 168 refs. Medical physics of x-ray contrast, metals, metal salts, and metal particulates, metal cluster complexes, organobismuth compds., and heavy metal complexes as x-ray contrast agents are discussed.

SUPPL. TERM: review metal x ray contrast media  
INDEX TERM: Imaging agents  
(contrast, radiog.; metal-based x-ray contrast media)  
INDEX TERM: Metals, biological studies  
ROLE: THU (Therapeutic use); BIOL (Biological study); USES  
(Uses)

- (metal-based x-ray contrast media)
- REFERENCE COUNT: 226 THERE ARE 226 CITED REFERENCES AVAILABLE FOR THIS RECORD.
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L22 ANSWER 9 OF 9 CAPLUS COPYRIGHT 2009 ACS on STN  
 ACCESSION NUMBER: 1999:252555 CAPLUS  
 DOCUMENT NUMBER: 131:129767  
 ENTRY DATE: Entered STN: 26 Apr 1999  
 TITLE: Secondary metabolite biosynthesis: the first century  
 AUTHOR(S): Bentley, Ronald  
 CORPORATE SOURCE: Department of Biological Sciences, University of  
     Pittsburgh, Pittsburgh, PA, 15260, USA  
 SOURCE: Critical Reviews in Biotechnology (1999), 19(1), 1-40  
 CODEN: CRBTE5; ISSN: 0738-8551  
 PUBLISHER: CRC Press LLC  
 DOCUMENT TYPE: Journal; General Review  
 LANGUAGE: English  
 CLASSIFICATION: 26-0 (Biomolecules and Their Synthetic Analogs)  
 Section cross-reference(s): 31, 34

**ABSTRACT:**  
 A review with 233 refs. on the biosynthesis of secondary metabolites up to 1965 is presented. The earliest suggestions for three of the four major pathways were speculative; for the isoprene rule, hypotheses date to 1877, for the polyketide rule to 1907, and for a role for amino acids in alkaloid biosynthesis to 1910. The fourth major pathway based on intermediates of the shikimic acid pathway has a much shorter history because shikimic acid itself was only identified as a primary metabolite in 1951. In addition to speculation, biomimetic syntheses were carried out in which chemists attempted to duplicate possible biosynthetic pathways *in vitro*. The classic example was Robinson's synthesis of tropinone in 1917. Direct examination of secondary metabolite biosynthesis was possible with the use of the isotopic tracer technique. This methodol., applied extensively to primary metabolism beginning in 1935 and to secondary metabolism from about 1950, was facilitated by the increasing availability of the <sup>14</sup>C isotope. With the use of isotopes as tracers, the broad outlines of secondary metabolite biosynthesis, reviewed here, were established in the period 1950 to 1965.

SUPPL. TERM: review secondary metabolite biosynthesis metabolic pathway  
 INDEX TERM: Metabolism  
     (anabolic; major pathways in the biosynthesis of  
     secondary metabolites)  
 INDEX TERM: Synthesis  
     (biomimetic; major pathways in the biosynthesis of  
     secondary metabolites of)  
 INDEX TERM: Metabolic pathways  
     (major pathways in the biosynthesis of secondary  
     metabolites)  
 INDEX TERM: Alkaloids, preparation  
 Polyketides  
 ROLE: BPN (Biosynthetic preparation); BIOL (Biological  
     study); PREP (Preparation)  
     (major pathways in the biosynthesis of secondary  
     metabolites of)  
 INDEX TERM: Radionuclides, preparation  
 ROLE: BSU (Biological study, unclassified); SPN (Synthetic  
     preparation); BIOL (Biological study); PREP (Preparation)  
     (major pathways in the biosynthesis of secondary  
     metabolites of)  
 INDEX TERM: Amino acids, reactions  
 ROLE: RCT (Reactant); RACT (Reactant or reagent)  
     (major pathways in the biosynthesis of secondary

INDEX TERM: metabolites of)  
Metabolic pathways  
(shikimate pathway; major pathways in the biosynthesis of secondary metabolites of)

INDEX TERM: 138-59-0P, Shikimic acid  
ROLE: BPN (Biosynthetic preparation); BSU (Biological study, unclassified); MFM (Metabolic formation); RCT (Reactant); BIOL (Biological study); FORM (Formation, nonpreparative); PREP (Preparation); RACT (Reactant or reagent)  
(major pathways in the biosynthesis of secondary metabolites of)

INDEX TERM: 532-24-1P, Tropinone  
ROLE: BPN (Biosynthetic preparation); SPN (Synthetic preparation); BIOL (Biological study); PREP (Preparation)  
(major pathways in the biosynthesis of secondary metabolites of)

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